

RESEARCH PAPER

# **Determination of Selected Metal Concentrations in Arable Soil Samples of Lapai Using Thermal Neutron Activation Analysis Technique**

**1 Ibrahim, A.A., <sup>2</sup>Oladipo, M.A.O., <sup>1</sup>Abdulkarim, A. and <sup>3</sup> Ibrahim, U.F.**

*<sup>1</sup>Department of Chemistry, Ibrahim Badamasi Babangida University, Lapai, Nigeria <sup>2</sup>Centre for Energy Research and Training (CERT), Ahmadu Bello University, Zaria <sup>3</sup>Department of Biotechnology, Nile University, Abuja*

# **ABSTRACT**

Soil is an important and abundant natural resource; it is also an important aspect of the environment as it serves as a sink for by-products of natural and anthropogenic processes. Plants obtain essential nutrients for growth from soil, other chemically similar elements that exist in soil compete for uptake by plant. Thus, these toxic substances accumulate in plants and transfer along food chain. Analysis of soil from selected locations of arable soil of Lapai was carried out using thermal neutrons activation analysis technique (TNAA) produced from Nigeria Research Reactor -1 (NIRR-1) operated at a power of 31kW and a neuron flux of 2.50 x 1011 cm-2s-1. Seven soil samples were collected from three sites, four elements were determined in all the samples collected. The elements determined were chromium, iron, cobalt and arsenic. The concentrations of all elements found were within permissible limit for agricultural activities.

**Keywords**: Arable soils, thermal neutron, TNAA Technique, MNSR.

# **INTRODUCTION**

Soil is an important part of the earth crust as it serves as a medium for plant growth as it contains essential nutrients for proper growth. It also serves as habitat for microorganisms and as an ultimate sink for by-products of natural and anthropogenic processes (Chatchai *et al*. 2019). These has made soil a complex system. Activities of man such as fossil fuel combustion, mining and indiscriminate refuse and sewage disposal has continue to alter the balance of nutrients present in soil that are required for proper plant growth (Lago-Vila *et al* 2015). these processes have led to lingering cases of environmental deterioration by increasing toxic substances above permissible level. Use of pesticides, fungicides and fertilizers have been reported as the primary anthropogenic source of trace metals in agricultural soil (Kelepertzis, 2014). The effects of these metals on plants range from mild to severe toxicity.

Trace elements concentrations such as Co, Fe, As and Cr have increased slightly in areas where varying pollution have been reported (Khan *et al*, 2008). Increase in abundance of these elements interferes with availability of essential nutrient required for proper plant growth (Gopal *et al*. 2003). More so, bioaccumulation of these chemicals along food chain pose a serious concern to human (Zhuang *et al*.2008). Some of the effects of these substances to man include damage to skin and mucous membrane, pulmonary and perforation of nasal septum. It has also been suggested that these substances are carcinogenic (Costa & Klein, 2006). A number of soil analysis have been carried out to determine concentration of these metals such as atomic absorption spectroscopy and recently inductive coupled plasma spectroscopy (Butler *et al*. 2015). These methods are destructive methods because samples will require digestion prior to analysis. More so, ICP is not readily available for use. Neutron activation analysis is a non-destructive method that is used for analysis elements of interest without matrices interference (Jonah *et al.* 2006 & Oladipo *et al.* 2012). This work was aimed at determining some selected metals in agricultural soil collected from some locations of Lapai in Savannah vegetation belt of Nigeria.

Received 4 December, 2020 Accepted 19 February, 2021 Address Correspondence to:

#### **METHODOLOGY**

#### **Sample Collection**

A comprehensive representative soils samples were collected from three locations of Lapai, Niger State. The sample locations with ordinance grid reference are Cece Village 09<sup>0</sup> 04'53N and 006°33' 42E Gulu 09° 02'44N and 006°36' 16E and IBB university farm 09° 04'42N and 006°34' 33E. The soil samples were collected at certain specified depths base on observable soil layer. An auger was used for digging to obtain loose soil samples. The samples were screened and stored in plastic bags. Table 2.1 shows the sample with corresponding horizon height and locations.

Table 2.1: Sampling Sites with their Locations and their Horizons



## **Sample Preparation**

Samples were prepared based on the procedure described by Oladipo *et al* (2012) After sampling on the field.

### **Experimental Analysis**

#### Measurement of pH

pH meter was calibrated with NIST Agros Organics buffer solutions of pH 4, 7 and 11. pH of samples were then measured by weighing 30 g of soil in 60 ml of water. The electrode was

immersed in flask and reading was taken after 5 minutes. The pH and the temperature of the solution were measured simultaneously and the results shown in Table 2.

#### Neutron Activation Analysis

A four-digit Melter model weighing balance was used for weighing samples. About 150 mg of soil was weighed, encapsulated and sealed in a polyethylene material. The capsules were packed into a cylindrical polyethylene vials and labelled accordingly. The samples were then injected into nuclear reactor.

#### Sample Irradiation

Miniature Neutron Source Reactor (MNSR) was used for irradiation of samples according to scheme described by Oladipo *et al*., (2012). The reactor was operated at full power of 30 kW and thermal neutron flux of  $2.50 \times 10^{11}$  ncm- $2s^{-1}$ .

### Gamma ray counting

High purity germanium (HPGe) detector was used for counting and acquisition of gamma ray. The HPGe detector was calibrated using standard radioactive sources of 137Cs (661.6 keV) and 60Co (1173 keV and 1332.5 keV). Gamma ray spectra obtained were processed using WINSPAN 2004 spectrum analysis software. The scheme for counting was followed as described by Oladipo *et al*., 2012).

#### **RESULTS AND DISCUSSION**

The pH of soils affects the availability of nutrients for plant uptake; therefore, pH is an important parameter to be determined in a soil analysis. The results of the pH readings of the savannah and rainforest regions of Nigeria are shown in Table 3.1.





*Concentrations are in ppm except otherwise stated*

The results of pH of samples are generally neutral and increase slightly with increasing depth height at sampled location. The range of pH found fall within 5.5 to 7.0 required for optimal availability of nutrients for plant growth as reported by Doerr (2007).

Assessment of results obtained in this study were carried out by analysing Coal fly ash standard reference material under same conditions used for determination of elements of interest in the samples. The concentrations of analytes of interest in reference material obtained using NIRR-1 (this work) were compared with values quoted on the reference certificate as given Table 3.2. A p value of 0.21 at α of 0.05 was obtained when t-Test was conducted. This suggests that the values obtained using NIRR-1 were not different from the true values of analytes.

Table 3.2: Results of elements concentrations in standard

	Coal fly ash	
Elements	This work	Reference value
Cr	$192.0 \pm 2.3$	$198.2 \pm 0.05$
Fe $(\% )$	$8.48 \pm 0.42$	$7.78 \pm 0.04$
Co	$50.8 \pm 0.09$	$50.1 \pm 0.01$
As	$135 \pm 1.90$	$136.2 \pm 0.8$
$\sigma$ , and the set of the set of $\sigma$ and $\sigma$ and $\sigma$ and $\sigma$ and $\sigma$		

*Concentrations are in ppm except otherwise stated*

Cr concentration range from  $16.2 \pm 0.1$  (Cece 1) to  $40.6 \pm 0.5$  ppm (IBBU farm). These falls within the range of 10 to 150 ppm reported in worlds soil (Zayed and Terry, 200). A general trend of chromium concentration observed was a decrease with increasing depth height in all the sampled locations except at Cece where only slight difference in concentrations was observed. The values of chromium of found in the sampled falls within the natural range of 10 to 50 ppm reported for natural occurrence in soil. Jain *et al*. reported decrease in germination of sugarbud in Cr concentration of 20 to 80 ppm (Shanker *et al*, 2005). It shows that the soil will not be suitable for cultivation of sugar cane at Gulu and IBBU farms while moderate concentration of Cr at Cece soil will support germination sugarcane buds. Cr level of 10 ppm causes reduction in dry plant weight. Other plants such as maize and tomato will only be affected by Cr toxicity above. However, plants commonly grown in the study such as maize, spinach, cabbage, sorghum, tomato, rice, and soya beans have been reported to increase uptake and accumulation with increasing soil Cr concentration (Jain *et al*., 2000; Shanker *et al*, 2005). Cr also interferes with uptake of nutritionally essential elements such as Fe, K, P Mg and Mn in soybeans root due to its structural similarity, therefore causing Fe deficiency in plants.

Fe concentration observed range from  $0.420 \pm$ 0.006 % (Cece 2) to  $1.38 \pm 0.014$  % (IBBU farm). concentration trend observed showed a decrease in concentration with increasing depth height. However, at Gulu top horizon gave slightly lower concentration than lower horizon. The samples contain high amount Fe. However, these concentrations may not be toxic to plants since not all of these Fe will not be available to plant for uptake because the pH is not acidic (below pH of 5). In addition, the soil may not be deficient in available Fe because the soil is not alkaline or basic as suggested by Anumalla *et al*. (2019).

Concentration range from  $1.87 \pm 0.03$  ppm (Cece 2) to  $7.43 \pm 0.06$  ppm (Gulu 2). variation was observed with increase depth height at Cece sampling site while increase in concentration was observed at Gulu with increasing sampling site. Slight increment was observed in samples collected at IBBU farm. It has been argued that excess amount of Co affects translocation of Fe. It has also been reported that increase in Co causes decrease in biomass and restricts uptake of essential nutrient (Gopal *et al*. 2003).

The range of concentration of As in the sampled locations is  $0.351 \pm 0.008$  ppm to  $0.942 \pm 0.006$ ppm. Concentration was observed to decrease generally with increase in horizon depth. The range is very low compared with permissible level by US environmental agency of 24 ppm and about 2 ppm found in bioaccessible part of United Kingdom (Palumbo-Roe *et al*., 2005). Plants such as rice have been reported to accumulate large percentage of arsenic. FAO has recommended 1 ppm as the permissible level As in rice. The sample locations contained low concentration of As which originated from sources such as use of herbicides, insecticides and phosphate fertilizer. This also explains the erratic trend in concentration with varying horizon height and sample location

#### **Conclusion**

Thermal neutron activation analytical technique was used to determine concentrations of Cr, Co, As and Fe in arable soil of Lapai. The results showed that concentrations of element were within permissible limit for agriculture activities such as farming in all the samples locations. The results also suggest that TNAA was successfully used to determine concentration of elements in soil samples.

### **REFERENCES**

- Anumalla, M. B. P., Mallikarjuna, S., Annamalai, A., & Jauhar, A. (2019). Tolerance of Iron-Deficient and -Toxic Soil Conditions in Rice. *Plants 8,* 1-34.
- Chatchai, S., Sakunna, W., Chanida, K., Choochad, S., Sila, K., (2019). Investigation of mobility and bioavailability of arsenic in agricultural soil after treatment by various soil amendments using sequential extraction procedure and multivariate analysis. *Catena, 181,* 1-11.
- Costa, M., & Klein, B. C. (2006). Toxicity and Carcinogenicity of Chromium Compounds in Humans. *Critical Reviews in Toxicology, 36*, 155-163.
- Doerr, S. H, Ritsema, C.J., Dekker, L.W., Scott, & Carter, D. (2007). Water repellence of soils: new insights and emerging research needs. *Hydrological processes, 21,* 222-2228.
- Gopal, R., Dube, B. K., Sinha, P., and Chatterjee, C. (2003). Cobalt Toxicity Effects on Growth and Metabolism of Tomato. *Communications in Soil Science and Plant Analysis, 34*, 619 — 628.
- Jain, R., Srivastava, S., Solomon, S.[,](https://link.springer.com/article/10.1007/s11738-010-0487-9#auth-4) [Shrivastava](https://link.springer.com/article/10.1007/s11738-010-0487-9#auth-4) A.K & [Chandra](https://link.springer.com/article/10.1007/s11738-010-0487-9#auth-5) A. (2010). Impact of excess zinc on growth parameters, cell division, nutrient accumulation, photosynthetic pigments and oxidative stress of sugarcane (*Saccharum* spp*.*) *Acta Physiol Plant*, 32**,** 979–986.
- Jonah, S. A, Umar, I. M., Oladipo, M. A. O., Balogun I. G., & Adeyamo, D. J. (2006). Standardisation of NIRR-1 Irradiation and counting facilities for Instrumental Neutron Activation analysis. *Journal of Applied Radiation and Isotopes*, *64*, 818-822.
- Karen, J. M., Mylene L. M., & Bradley, M. T. (2005). Cr(III) Oxidation and Cr Toxicity in Cultures of the Manganese(II)-Oxidizing *Pseudomonas putida* Strain GB-1. *Geomicrobiology Journal*, *22*, 151–159.
- Kelepertzis E. (2014). Accumulation of heavy metals in agricultural soils of Mediterranean: Insights from Argolida Basin, Peloponnese, Greece. *Geoderma, 221*, 82-90.
- Khan, S., Cao, Q., Zheng, Y. M., Huang, Y.Z., & Zhu Y.G. (2008). Health risks of heavy metals in contaminated soils and food crops irrigated with wastewater in Beijing, China. *Environmental Pollution*, 686-692.
- Lago-Vila, M., Arenas-Lago, D., Rodríguez-Seijo, A., Andrade, C. M.L., & Vega F.A, (2015). Cobalt, chromium and nickel contents in soils and plants from a serpentinite quarry, *Solid Earth, 6,* 323–335.
- Liu, Y., Wang, H., Li, X., & Li J. (2015). Heavy metal contamination of Agricultural soils in Taiyuan, China. *Pedosphere 25,* 901-909.
- Mico, C., Recatala, L., Peris, M., & Sa´nchez, J. (2006). Assessing heavy metal sources in agricultural soils of an European Mediterranean area by multivariate analysis. *Chemosphere 65,* 863–872.
- Njinga, R.L., Jonah, S.A., Ewa, I.O.B., Oladipo, M.O.A., & Agbo, G.A. (2011). K0-IAEA determination of full energy peak efficiency for a high purity germanium detector. *Indian Journal Science Research, 3*, 25-32.
- Owen, T. B., Warren, R. L. C., Jennifer, M. C., & Christine, M. D. (2015). 2014 Atomic spectrometry update – a review of advances in environmental analysis. *Journal of Analytical Atomic Spectrom, 30,* 21–63.
- [Shanker,](https://www.sciencedirect.com/science/article/pii/S0160412005000231#!) A. K[, Cervantes](https://www.sciencedirect.com/science/article/pii/S0160412005000231#!) C., [Loza-Taverac,](https://www.sciencedirect.com/science/article/pii/S0160412005000231#!) H. & [Avudainayagam, S.](https://www.sciencedirect.com/science/article/pii/S0160412005000231#!) (2005). Chromium toxicity in plants, *[Environment International](https://www.sciencedirect.com/science/journal/01604120)[,](https://www.sciencedirect.com/science/journal/01604120/31/5)  [31,](https://www.sciencedirect.com/science/journal/01604120/31/5)* 739-753.
- Zhuang, P., Zou, H., Shu, W., (2009). Biotransfer of heavy metals along a soil-plantinsect-chicken food chain: Field study. *Journal of Environmental Sciences, 21,* 849– 853.